

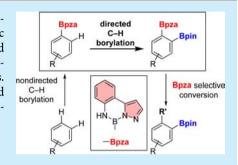
# Regioselective Synthesis of *o*-Benzenediboronic Acids via Ir-Catalyzed *o*-C—H Borylation Directed by a Pyrazolylaniline-Modified Boronyl Group

Takeshi Yamamoto, Aoi Ishibashi, and Michinori Suginome\*

Department of Synthetic Chemistry and Biological Chemistry, Graduate School of Engineering, Kyoto University, Katsura, Nishikyo-ku, Kyoto 615-8510, Japan

Supporting Information

**ABSTRACT:** Ir-catalyzed *ortho*-directed C–H borylation of pyrazolylaniline (PZA)-modified arylboronic acids with bis(pinacolate)diboron afforded *o*-benzenediboronic acids in which two boronyl groups are differentially modified by pinacol (PIN) and PZA. By using this borylation after nondirected Ir-catalyzed C–H borylation, *o*-benzenediboronic acids are conveniently synthesized from unfunctionalized arenes. The differentially modified *o*-benzenediboronic acids undergo selective oxidation and Suzuki–Miyaura cross-coupling at the PZA-modified boronyl groups, affording *o*-functionalized arylboronic acids selectively.



o-Benzenediboronic acid is an attractive structural motif not only as a building block and synthetic intermediate for the synthesis of natural products, <sup>1</sup> polycyclic aromatic hydrocarbons, <sup>2</sup> and oligoarene derivatives <sup>3</sup> but also as a bidentate Lewis acid <sup>4</sup> and key organic component of hydrogen-bonded organic frameworks. <sup>5</sup> In view of the rapid expansion of the utilization of functionalized boronic acid derivatives in various research fields including material science and organocatalysis, it is important to develop efficient and convenient synthetic access to σ-benzenediboronic acids.

The synthesis of *o*-benzenediboronic acids **A** has generally relied on the use of 1,2-difunctionalized benzene derivatives **B** as starting materials (Scheme 1). In typical cases, *o*-halobenzeneboronic acids or 1,2-dihalobenzenes are converted into *o*-benzenediboronic acids through halogen—metal (Li or Mg) exchange 5,6d-g or palladium-catalyzed Miyaura borylation. 1,2b,c,3a,7 Recently, catalytic diborylation of arynes generated from 1,2-difunctionalized benzene derivatives has

Scheme 1. Preparation of o-Benzenediboronic Acids

attracted much attention as a convenient route to obenzenediboronic acids. https://display.com/specificient-and-selective, the requirement for 1,2-difunctionalized arenes as starting material detracts much from the synthetic efficacy of the approach. Notably, high-yielding, photoinduced dual C-H/C-X borylation of monohaloarenes C has been reported quite recently. However, although predictable, the selectivity for the formation of 1,2-diborylated products over 1,3-diborylated products depends to a significant degree on the electronic effect of substituents on the aromatic group. Furthermore, it is highly desirable to utilize simple arylboronic acids D or even nonfunctionalized arenes E as starting materials in the synthesis of o-benzenediboronic acids through C-H activation without using functionalized arenes bearing halogen or pseudohalogen leaving groups.

We previously reported Ru-catalyzed *o*-C–H silylation of arylboronic acids by attaching a temporary directing group on the boron atom of arylboronic acids. Modification of the boronyl group with pyrazolylaniline or anthranilamide, which leads to the formation of the Bpza or Baam group, respectively, was found to be critical for the success of the highly selective directed transformation. In this paper, we describe the use of the PZA-modified boronyl groups as a temporary directing group in Ir-catalyzed *o*-C–H borylation. Although *o*-C–H borylation directed by several heteroatom functional groups was achieved by using Ir<sup>13</sup> and other transition-metal catalysts, <sup>14</sup> to our knowledge, boronyl-group-directed *o*-C–H borylation has not been reported. By utilizing the directed borylation, nonfunctionalized arenes are transformed into 1,2-

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benzenediboronic acid derivatives through nondirected C-H borylation followed by Bpza-directed *o*-C-H borylation.

PZA-modified phenylboronic acid **1a**, synthesized by condensation of phenylboronic acid and 2-pyrazol-3-ylaniline, <sup>12a</sup> reacted with diboron in the presence of catalytic amounts of transition-metal complexes (Table 1). The resultant reaction

Table 1. o-C-H Functionalization of Phenylboronic Acid Derivatives $^a$ 

entry	substrate (1)	catalyst	solvent	yield (%) of 3a
1	PhBpza (1a)	[Ir(OMe)(cod)] <sub>2</sub> (4)	THF	85
2	1a	4, dtbpy	THF	<1 (15) <sup>c</sup>
3	1a	$[IrCl(cod)]_2$	THF	59
4	1a	$[RhCl(cod)]_2$	THF	5
5	1a	$[RhOH(cod)]_2$	THF	2
6	1a	4	cyclohexane	60
7	1a	4	toluene	57
8	1a	4	1,4-dioxane	67
9	1a	4	1,2-DME <sup>d</sup>	68
10 <sup>e</sup>	1a	4	THF	74
11	$PhB(OH)_2$	4	THF	0
12	PhBpin	4	THF	0
13	PhBdan	4	THF	0
14	PhBaam	4	THF	0
15 <sup>f</sup>	1a	4	THF	70
$16^{e,f}$	1a	4	THF	86

<sup>a</sup>Reaction conditions: 1 (0.10 mmol), B<sub>2</sub>pin<sub>2</sub> (0.15 mmol), and catalyst (1.0  $\mu$ mol), THF (1.0 mL), 80 °C, 18 h. <sup>b</sup>GC yield. <sup>c</sup>NMR yield of m- and p-diborylbenzenes. <sup>d</sup>1,2-Dimethoxyethane. <sup>e</sup>Reaction at 110 °C. <sup>f</sup>HBpin (0.30 mmol) instead of B<sub>2</sub>pin<sub>2</sub>.

mixture was treated with pinacol for detection and quantification of borylation product 2 in the form of pinacol ester (3a). In the presence of 1.0 mol % [Ir(OMe)(cod)]<sub>2</sub>, o-C-H borylation proceeded at 80 °C for 18 h, giving 3a in 85% yield (entry 1). Notably, unreacted 1a and double borylated product were recovered as PhBpin (10%) and 1,2,3-benzenetriboronic acid pinacol ester (5a, 5%), respectively. Addition of 4,4'-di-tert-butyl-2,2'-bipyridyl (dtbpy), which is recognized as the most effective ligand in nondirected C-H borylation, resulted in m- or p-C-H borylation preferentially (15%, m:p =1:2) (entry 2). The reaction with [IrCl(cod)]<sub>2</sub> afforded 3a in 59% yield (entry 3). Although [RhCl(cod)]<sub>2</sub> and [RhOH-(cod)]2 showed low catalyst activities, C-H borylation proceeded at the o-position selectively (entries 4 and 5). The use of cyclohexane, toluene, 1,4-dioxane, or 1,2-dimethoxyethane as solvent resulted in lower yields, with recovery of unreacted 1a as PhBpin (entries 6–9). Performing the reaction at 110 °C gave 3a in lower yield and with a larger amount of 5a

(14%) (entry 10). The requirement for the *ortho*-directing group was confirmed by the observation that no product was formed in the reactions of PhB(OH)<sub>2</sub> or PhBpin under the same reaction conditions (entries 11 and 12). Furthermore, the use of PhBdan bearing a 1,8-diaminonaphthalene (DAN)-masked boronyl group afforded no desired product (entry 13). PhBaam underwent C–H borylation mainly at the anthranila-mide group rather than at the *o*-C–H bond (entry 14). It was also found that use of HBpin instead of B<sub>2</sub>pin<sub>2</sub> as a boron source successfully gave the *o*-borylation product in good yield (entry 15). In this case, a higher yield was obtained by raising the reaction temperature to 110 °C (entry 16). The primary product 2, bearing Bpza and Bpin groups, was isolated by crystallization, albeit in moderate yield, and found to be hydrolytically unstable.

Under the optimized reaction conditions, various PZA-modified arylboronic acids were subjected to the directed C-H borylation (Scheme 2). All the products were isolated after

Scheme 2. Substrate Scope a,b

"Reaction conditions: 1 (0.30 mmol),  $B_2pin_2$  (0.45 mmol),  $[Ir(OMe)(cod)]_2$  (3.0  $\mu$ mol), THF (3.0 mL), 80 °C, 18 h. <sup>b</sup>NMR yield. Isolated yields in parentheses. <sup>c</sup> $[Ir(OMe)(cod)]_2$  (7.5  $\mu$ mol).

conversion into the corresponding bispinacolates after treatment of the reaction mixtures with pinacol. The reaction of *p*-substituted compounds bearing methoxy, trifluoromethyl, and methyl groups afforded 4-substituted 1,2-diborylbenzenes 3b—d. The directed C—H borylation of *m*-substituted arenes proceeded at the less hindered *o*-positions, giving 4-substituted 1,2-diborylbenzenes 3b—g selectively. *o*-Substituted compounds could also be employed in the C—H borylation, giving the corresponding 3-substituted 1,2-benzenediboronates 3h—j in high yields. 1-Naphthyl- and 2-naphthylboronic acids gave 1,2- and 2,3-diborylnaphthalenes 3k and 3l, respectively, in high yield and with excellent selectivity.

The Bpza-directed *o*-borylation, in combination with non-directed C–H borylation, <sup>15</sup> allowed the synthesis of 1,2-benzenediboronic acids from nonfunctionalized arenes (Scheme 3). 1-Chloro-2-methoxybenzene was subjected to Ircatalyzed C–H borylation with B<sub>2</sub>pin<sub>2</sub> in the presence of [Ir(OMe)(cod)]<sub>2</sub>, giving a mixture of two regioisomers in a 1:1 ratio. <sup>16</sup> The mixture was treated with 2-pyrazol-3-ylaniline and subjected to the directed C–H borylation under the standard reaction conditions. After treatment with pinacol, 3-chloro-4-

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# Scheme 3. Synthesis of o-Benzenediboronic Acid Derivatives from Nonfunctionalized Arenes<sup>a</sup>

"Reaction conditions: (a)  $[Ir(OMe)(cod)]_2$  (1.5 mol %), dtbpy (3.0 mol %),  $B_2pin_2$  (1.2 equiv), THF, rt, 18 h; (b) NaIO<sub>4</sub> (3 equiv), THF, H<sub>2</sub>O, 30 min, then HCl aq, 4 h; (c) H<sub>2</sub>pza, toluene, reflux; (d)  $[Ir(OMe)(cod)]_2$  (1.0 mol %),  $B_2pin_2$  (1.5 equiv), THF, 80 °C, 18 h; (e) pinacol (2.0 equiv), PTSA·H<sub>2</sub>O (2.0 equiv), THF, 80 °C, 1 h; (f)  $[Ir(OMe)(cod)]_2$  (0.5 mol %), dtbpy (1.0 mol %),  $B_2pin_2$  (1.0 equiv), toluene (30 equiv), rt, 30 h; (g)  $[Ir(OMe)(cod)]_2$  (2.0 mol %),  $B_2pin_2$  (1.5 equiv), THF, 80 °C, 18 h.

methoxy-1,2-benzenediboronate (3m) was isolated in good overall yield as a single regioisomer. This four-step protocol was also applicable to monosubstituted benzenes, as demonstrated by the conversion of toluene into 4-methyl-1,2-benzenediboronic acid (3d) (Scheme 2).

Our attention then turned to the reactivities of **2**, bearing two different masking groups, namely, PIN and PZA, on the boron atoms. When *m*-TolBpza (*m*-1d) was subjected to the directed C—H borylation and subsequent oxidation with trimethylamine oxide (TMAO, 2.5 equiv), we observed selective oxidation of one of the two different boron functional groups. The reaction gave *o*-hydroxyarylboronic acid **6**, which has been reported to be an important coupling building block in iterative cross-coupling for the synthesis of oligoarene derivatives, in a highly chemoselective manner (Scheme 4).<sup>17,18</sup> It is interesting to

Scheme 4. One-Pot Synthesis of *o*-Hydroxyarylboronic Acid Derivatives

note that the Bpza group was oxidized selectively over the Bpin group. When *m*-TolBpza and *m*-TolBpin were separately treated with TMAO, we found that Bpza was more prone to oxidation (see Supporting Information). Use of regioisomeric TolBpza as a starting material allowed the synthesis of regioisomeric products **6a** and **6c** with high selectivity. Similarly, a pair of regioisomeric methoxy-substituted *o*-hydroxyarylboronic acid derivatives (**6b** and **6d**) was prepared selectively.

The Bpza selective conversion was also observed in Suzuki–Miyaura cross-coupling (Scheme 5). After the directed *o*-C–H

Scheme 5. One-Pot Synthesis of o-Borylated Biaryls

borylation of *p*-TolBpza (*p*-1d) using 3 equiv of HBpin, the reaction mixture was treated with isopropyl alcohol to decompose the remaining HBpin. Suzuki—Miyaura crosscoupling of the obtained crude product bearing Bpza and Bpin groups with 1-bromonaphthalene or ethyl 2-bromobenzoate proceeded at the Bpza group with high selectivity, giving biaryls 7a and 8a, bearing a Bpin group, along with the formation of a small amount of regioisomers 7b and 8b. Under the same reaction conditions, *m*-TolBpza (*m*-1d) was converted into methyl-substituted biaryls 7b and 8b, bearing a Bpin group, with high regioselectivity, in one-pot.

In conclusion, we have demonstrated a new Ir-catalyzed directed C–H borylation of arylboronic acid derivatives by attaching a removable directing group on the boron atoms. By combining with the nondirected C–H borylation, the reaction provides a new, convenient access to *o*-benzenediboronic acids from nonfunctionalized arenes through conversion of the two adjacent aromatic C–H bonds on the aromatic rings. The unsymmetrically modified *o*-benzenediboronic acids with PZA and PIN undergo selective conversion at the Bpza group in the TMAO oxidation and Suzuki–Miyaura cross-coupling.

#### ASSOCIATED CONTENT

# Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.7b00041.

Detailed experimental procedures and compound characterization data (PDF)

# AUTHOR INFORMATION

#### **Corresponding Author**

\*E-mail: suginome@sbchem.kyoto-u.ac.jp.

# ORCID ®

Michinori Suginome: 0000-0003-3023-2219

# Notes

The authors declare no competing financial interest.

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